## Section 3.3. Rate Problems

Note. In this section we consider three types of rate problems: (1) Decay Problems, (2) Growth Problems, and (3) Mixture Problems.

Note. The rate at which radioactive nuclei decay is proportional to the number of such nuclei present. With $x$ as the mass of radioactive substance, we have the DE $\frac{d x}{d t}=-k x$ where $k>0$. Suppose $x(0)=x_{0}$. Then $x=x_{0} e^{-k t}$.

Definition. The half life of a radioactive substance is the time $T$ it takes for the amount of substance present to reach $1 / 2$ the original amount. So we set $\frac{1}{2} x_{0}=x_{0} e^{-k T}$ so that $T=\ln (2) / k$. The half life of some common radioactive elements is as follows.

| Isotope | Half Life |
| :---: | :---: |
| Uranium, $\mathrm{U}^{238}$ | $4.5 \times 10^{9}$ years |
| Plutonium, $\mathrm{Pu}^{230}$ | 24,360 years |
| Carbon, $\mathrm{C}^{14}$ | 5,730 years |
| Radium, $\mathrm{Ra}^{226}$ | 1,620 years |
| Einsteinium, Es ${ }^{254}$ | 276 days |
| Nobelium, $\mathrm{No}^{257}$ | 23 seconds |

Example. Carbon 14 dating was performed on the Shroud of Turin in late 1989. The organic fibers of the clothe were found to contains $89 \%$ of the $\mathrm{C}^{14}$ of such fibers today. How old is the shroud?

Solution. The amount of $\mathrm{C}^{14}$ present at a time $t$ after the decay starts (that is, after the plant producing the fibers has died) is $x=x_{0} e^{-k t}$. First, we need to find $k$. Since we have the half life of $\mathrm{C}^{14}$ above, we can set $x=x_{0} / 2$ and $t=5,730$ years to get $x_{0} / 2=x_{0} e^{-k(5,730 \text { years) }}$ and we find $k \approx 0.00012097$ (when $t$ is expressed in years). Now with the amount as $x=(0.89) x_{0}$ and this value of $k$, we can solve for age $t: 0.89 x_{0}=x_{0} e^{-(0.00012097) t}$ or $0.89=e^{-(0.00012097) t}$ or $t \approx 963$ years. Notice that this places the estimated date of the shroud as $1989-963=1026$.

Note. We now shift form exponential decay to exponential growth and study "Growth Problems." In an ideal environment, the rate of growth of a population is proportional to the size of the population, i.e. $\frac{d x}{d t}=k x$ where $k>0$ and so $x=x_{1} e^{k t}$. As with half life in exponential decay, we can talk about doubling time $T$ and get $T=\ln (2) / k$, as before.

Note. In the "real world," no environment is ideal. In fact, an environment will have some carrying capacity, $K$. This reflects the maximum number of individuals which an ecosystem can support. The limits are due to crowding, competition for food, mates, etc. As the size of the population nears $K$, the rate of growth of the population slows. This is modeled by the differential equation $\frac{d x}{d t}=k x(K-x)$. Notice for $x>K$ we have $d x / d t<0$ and the population decreases. We can also express the DE as

$$
\frac{d x}{d t}=k x-\lambda x 2=\lambda x\left(\frac{k}{\lambda}-x\right) .
$$

Growth which follows the DE is called logistic growth. Notice the DE is separable.

If $x(0)=x_{0}$ then

$$
x(t)=\frac{k x_{0}}{\lambda x_{0}+\left(k-\lambda x_{0}\right) e^{-k t}} .
$$

Notice that $\lim _{t \rightarrow \infty} x(t)=k / \lambda$; this is the carrying capacity in terms of $k$ and $\lambda$.

Note. We now consider the concentration of a substance (such as salt) in a tank. The tank will have the substance being added to the tank at some fixed rate, and material is removed from the tank at a fixed rate. We will assume the substance which is added to the tank is instantaneously mixed. The is an example of a "Mixture Problem."

Example. Page 105 Number 21. A tank initially contains 100 gal of brine in which there is dissolved 20 lb of salt. Starting at time $t=0$, brine containing 3 lb of dissolved salt per gallon flows into the tank at the rate of $4 \mathrm{gal} / \mathrm{min}$. The mixture is kept uniform by stirring and the well-stirred mixture simultaneously flows out of the tank at the same rate. Find the amount of salt in the tank as a function of time.

Solution. Let $x$ be the amount of salt at time $t$. Then $x_{0}=20 \mathrm{lb}$. The salt enters the tank at a rate of $(3 \mathrm{lb} / \mathrm{gal}) \times(4 \mathrm{gal} / \mathrm{min})=12 \mathrm{lb} / \mathrm{min}$. The salt leaves the tank at a rate of $(x \mathrm{lb} / \mathrm{gal}) \times(4 \mathrm{gal} / \mathrm{min})=x / 25 \mathrm{lb} / \mathrm{min}$. So $d x / d t=12-x / 25$ and $x(0)=20$. We have $25 \frac{d x}{d t}=300-x$ or $\frac{25}{300-x} d x=d t$ so that $-25 \ln |300-x|=$ $t+c$ for some $c$. This gives $\ln |300-x|=-t / 25-c / 25$ or (exponentiating) $300-x=e^{-t / 25} e^{-c / 25}=k e^{-t / 25}$ where $k=e^{-c / 25}$. So we have $x(t)=300-k e^{-t / 25}$. With $x(0)=20$ we find $k=280$ so that the amount of salt in the tank at time $t$ is $x(t)=300-280 e^{-t / 25}$.

