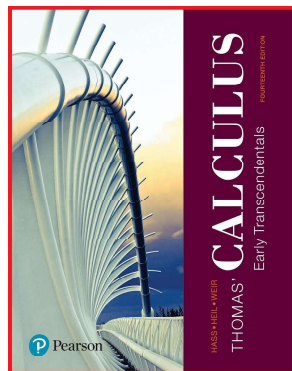


# Calculus 1

## Chapter 1. Functions

### 1.5. Exponential Functions—Examples and Proofs

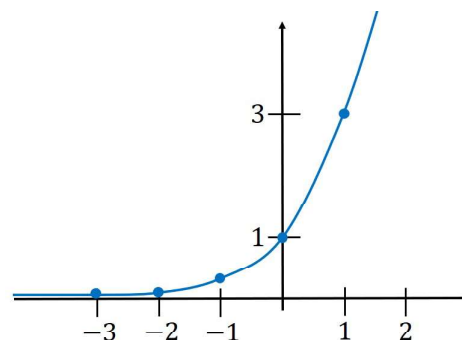


### Exercise 1.5.2(a)

**Exercise 1.5.2(a).** Plot several points and sketch the graph of  $y = 3^x$ .

**Solution.** Consider the function values:

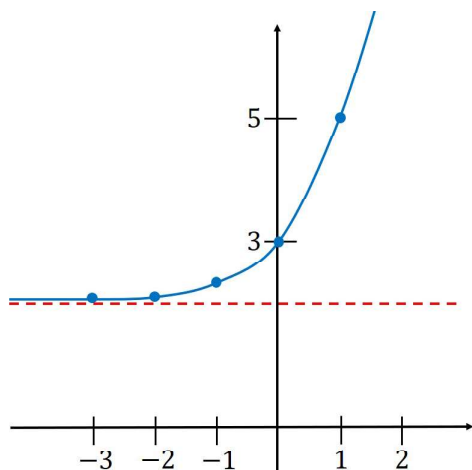
$x$	-3	-2	-1	0	1	2	3
$f(x)$	$3^{-3} = 1/27$	$3^{-2} = 1/9$	$3^{-1} = 1/3$	$3^0 = 1$	$3^1 = 3$	$3^2 = 9$	$3^3 = 27$



### Exercise 1.5.8(a)

**Exercise 1.5.8(a).** Sketch the shifted exponential curve  $y = 3^x + 2$ .

**Solution.** We simply shift the graph of  $y = 3^x$  from Exercise 1.5.2(a) up by 2 units (because of the “+2”). Notice that the graph of  $y = 3^x + 2$  has a horizontal asymptote of  $y = 2$ .



### Exercise 1.5.12

**Exercise 1.5.12.** Use the Law of Exponents (Theorem 1.5.A) to simplify  $9^{1/3}9^{1/6}$ .

**Solution.** We have by the Rules for Exponents (Theorem 1.5.A) that  $a^x a^y = a^{x+y}$ .

So with  $a = 9$ ,  $x = 1/3$ , and  $y = 1/6$ , we have  $9^{1/3}9^{1/6} = 9^{1/3+1/6} = 9^{1/2} = \boxed{3}$ .  $\square$

## Exercise 1.5.30(a)

**Exercise 1.5.30(a).** Population Growth.

The population of Silver Run in the year 1890 was 6250. Assume the population increased at a rate of 2.75% per year (that is,  $k = 0.0275$ ). **(a)** Estimate the population in 1915 and 1940.

**Solution.** The population size is given by  $y = y_0 e^{kt}$  where  $k = 0.0275$ ,  $y_0 = 6250$ , and  $t$  is time measured in years after 1890.

In 1915,  $t = 1915 - 1890 = 25$  and so the population is  $y = 6250e^{0.0275(25)} = 6250e^{0.6875} \approx \boxed{12,430}$ .

In 1940,  $t = 1940 - 1890 = 50$  and so the population is  $y = 6250e^{0.0275(50)} = 6250e^{1.375} \approx \boxed{24,719}$ .  $\square$

## Example 1.5.4

**Example 1.5.4.** Laboratory experiments indicate that some atoms emit a part of their mass as radiation, with the remainder of the atom reforming to make an atom of some new element. For example, radioactive carbon-14 decays into nitrogen; radium eventually decays into lead. If  $y_0$  is the number of radioactive nuclei present at time zero, the number still present at any later time  $t$  will be  $y = y_0 e^{-rt}$  where  $r > 0$ . The number  $r$  is the *decay rate* of the radioactive substance. For carbon-14, the decay rate has been determined experimentally to be about  $r = 1.2 \times 10^{-4}$  when  $t$  is measured in years. Predict the percent of carbon-14 present after 866 years have elapsed.

**Solution.** With  $r = 1.2 \times 10^{-4}$  and  $t = 866$ , we have  $u = y_0 e^{(-0.00012)(866)} = y_0 e^{-0.10392} \approx (0.9013)y_0$ . So the percent of carbon-14 present at this time is  $(0.9013y_0)/y_0 \times 100\% = \boxed{90.13\%}$ .