## Shepley L. Ross Introduction to Ordinary Differential Equations

## Chapter 3. Applications of First-Order Equations 3.3. Rate Problems

- **3.3.1.** Assume that the rate at which radioactive nuclei decay is proportional to the number of such nuclei that are present in a given sample. In a certain sample 10% of the original number of radioactive nuclei have undergone disintegration in a period of 100 years.
  - (a) What percentage of the original radioactive nuclei will remain after 1000 years?
  - (b) In how many years will only one-fourth of the original number remain?
- **3.3.5.** In a certain bacteria culture the rate of increase in the number of bacteria is proportional to the number present.
  - (a) If the number triples in 5 hours, how many will be present in 10 hours?
  - (b) When will the number present be 10 times the number initially present?
- **3.3.9.** A hot pie is taken directly from an oven and placed outdoors on a porch table to cool on a day when the surrounding outdoor temperature is a constant  $80^{\circ}$  F. The temperature of the pie was  $350^{\circ}$  F at the instant t = 0 when it is placed on the table, and it was  $300^{\circ}$  F 5 minutes later.
  - (a) What was the temperature 10 minutes after it was placed on the table?
  - (b) When was the temperature  $100^{\circ}$  F?
- **3.3.11.** Assume that the population of a certain city increases at a rate proportional to the number of inhabitants at any time. If the population doubles in 40 years, in how many years will it triple?
- **3.3.21.** A tank initially contains 100 gal of brine in which there is dissolved 20 lb of salt. Starting at time t = 0, brine containing 3 lb of dissolved salt per gallon flows into the tank at the rate of 4 gal/min. The mixture is kept uniform by stirring and the well-stirred mixture simultaneously flows out of the tank at the same rate.
  - (a) How much salt is in the tank at the end of 10 min?
  - (b) When is there 160 lb of salt in the tank?

**3.3.27(a).** The air in a room whose volume is  $10,000 \text{ ft}^3$  tests 0.15% carbon dioxide. Starting at t=0, outside air testing 0.05% carbon dioxide is admitted at the rate of  $5000 \text{ ft}^3/\text{min}$ . Assume this forces out well-mixed air from the room at the same rate. What is the percentage of carbon dioxide in the air in the room after 3 minutes?